

<p>Thallium (Tl) <b><u>atomic weight = 204</u></b> Oxygen combination: 2:3</p> <p>-----</p> <p><i>Transition Property</i> Boiling point of salt: 430</p>	<p>Sodium (Na) <b><u>atomic weight = 23</u></b> Oxygen combination: 2:1</p> <p>-----</p> <p><i>Transition Property</i> Electron affinity: 52</p>	<p>Rubidium (Rb) <b><u>atomic weight = 85</u></b> Oxygen combination: 2:1</p> <p>-----</p> <p><i>Transition Property</i> Electron affinity: 46</p>	<p>Boron (B) <b><u>atomic weight = 11</u></b> Oxygen combination: 2:3</p> <p>-----</p> <p><i>Transition Property</i> Boiling point of salt: 15</p>	<p>Indium (In) <b><u>atomic weight = 114</u></b> Oxygen combination: 2:3</p> <p>-----</p> <p><i>Transition Property</i> Boiling point of salt: 500</p>
<p>Yttrium (Y) <b><u>atomic weight = 88</u></b> Oxygen combination: 2:3</p> <p>-----</p> <p><i>Transition Property</i> Boiling point of salt: 1500</p>	<p>Oxygen (O) <b><u>atomic weight = 16</u></b> Oxygen combination: 1:3</p> <p>-----</p> <p><i>Transition Property</i> Electron affinity: 141</p>	<p>Gold (Au) <b><u>atomic weight = 197</u></b> Oxygen combination: 2:1</p> <p>-----</p> <p><i>Transition Property</i> Electron affinity: 222</p>	<p>Beryllium (Be) <b><u>atomic weight = 9</u></b> Oxygen combination: 1:1</p> <p>-----</p> <p><i>Transition Property</i> Electron affinity: -241</p>	<p>Fluorine (F) <b><u>atomic weight = 19</u></b> Oxygen combination: 2:7</p> <p>-----</p> <p><i>Transition Property</i> Electron affinity: 330</p>
<p>Mercury (Hg) <b><u>atomic weight = 200</u></b> Oxygen combination: 1:1</p> <p>-----</p> <p><i>Transition Property</i> Electron affinity: 0</p>	<p>Niobium (Nb) <b><u>atomic weight = 94</u></b> Oxygen combination: 2:5</p> <p>-----</p> <p><i>Transition Property</i> Melting Point: 2500</p>	<p>Antimony (Sb) <b><u>atomic weight = 122</u></b> Oxygen combination: 2:5</p> <p>-----</p> <p><i>Transition Property</i> Melting Point: 630</p>	<p>Lithium (Li) <b><u>atomic weight = 7</u></b> Oxygen combination: 2:1</p> <p>-----</p> <p><i>Transition Property</i> Electron affinity: 59</p>	<p>Cerium (Ce) <b><u>atomic weight = 140</u></b> Oxygen combination: 1:2</p> <p>-----</p> <p><i>Transition Property</i> Electron affinity: 0</p>
<p>Lead (Pb) <b><u>atomic weight = 207</u></b> Oxygen combination: 1:2</p> <p>-----</p> <p><i>Transition Property</i> Electron affinity: 106</p>	<p>Tantalum (Ta) <b><u>atomic weight = 182</u></b> Oxygen combination: 2:5</p> <p>-----</p> <p><i>Transition Property</i> Melting point: 3000</p>	<p>Tellurium (Te) <b><u>atomic weight = 128</u></b> Oxygen combination: 1:3</p> <p>-----</p> <p><i>Transition Property</i> Electron affinity: 190</p>	<p>Strontium (Sr) <b><u>atomic weight = 87</u></b> Oxygen combination: 1:1</p> <p>-----</p> <p><i>Transition Property</i> Electron affinity: -167</p>	<p>Iodine (I) <b><u>atomic weight = 127</u></b> Oxygen combination: 2:7</p> <p>-----</p> <p><i>Transition Property</i> Electron affinity: 300</p>
<p>Tungsten (W) <b><u>atomic weight = 184</u></b> Oxygen combination: 1:3</p> <p>-----</p> <p><i>Transition Property</i> Electron affinity: 58</p>	<p>Molybdenum (Mo) <b><u>atomic weight = 96</u></b> Oxygen combination: 1:3</p> <p>-----</p> <p><i>Transition Property</i> Electron affinity: 97</p>	<p>Bismuth (Bi) <b><u>atomic weight = 208</u></b> Oxygen combination: 2:5</p> <p>-----</p> <p><i>Transition Property</i> Melting point: 271</p>	<p>Zinc (Zn) <b><u>atomic weight = 65</u></b> Oxygen combination: 1:1</p> <p>-----</p> <p><i>Transition Property</i> Electron affinity: 0</p>	<p>Lanthanum (La) <b><u>atomic weight = 138</u></b> Oxygen combination: 2:3</p> <p>-----</p> <p><i>Transition Property</i> Boiling point of salt: 1800</p>

<p>Titanium (Ti) <b><u>atomic weight = 48</u></b> Oxygen combination: <b>1:2</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Electron affinity: 19</i></p>	<p>Vanadium (V) <b><u>atomic weight = 51</u></b> Oxygen combination: <b>2:5</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Melting point: 1890</i></p>	<p>Phosphorus (P) <b><u>atomic weight = 31</u></b> Oxygen combination: <b>2:5</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Melting point: 44</i></p>	<p>Calcium (Ca) <b><u>atomic weight = 40</u></b> Oxygen combination: <b>1:1</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Electron affinity: -156</i></p>	<p>Selenium (Se) <b><u>atomic weight = 78</u></b> Oxygen combination: <b>1:3</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Electron affinity: 194</i></p>
<p>Sulfur (S) <b><u>atomic weight = 32</u></b> Oxygen combination: <b>1:3</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Electron affinity: 200</i></p>	<p>Manganese (Mn) <b><u>atomic weight = 55</u></b> Oxygen combination: <b>2:7</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Electron affinity: 0</i></p>	<p>Chromium (Cr) <b><u>atomic weight = 52</u></b> Oxygen combination: <b>1:3</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Electron affinity: 64</i></p>	<p>Magnesium (Mg) <b><u>atomic weight = 24</u></b> Oxygen combination: <b>1:1</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Electron affinity: -232</i></p>	<p>Bromine (Br) <b><u>atomic weight = 80</u></b> Oxygen combination: <b>2:7</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Electron affinity: 324</i></p>
<p>Arsenic (As) <b><u>atomic weight = 75</u></b> Oxygen combination: <b>2:5</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Melting point: 817</i></p>	<p>Barium (Ba) <b><u>atomic weight = 137</u></b> Oxygen combination: <b>1:1</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Electron affinity: -52</i></p>	<p>Chlorine (Cl) <b><u>atomic weight = 36</u></b> Oxygen combination: <b>2:7</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Electron affinity: 350</i></p>	<p>Copper (Cu) <b><u>atomic weight = 63</u></b> Oxygen combination: <b>2:1</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Electron affinity: 118</i></p>	<p>Silver (Ag) <b><u>atomic weight = 108</u></b> Oxygen combination: <b>2:1</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Electron affinity: 125</i></p>
<p>Tin (Sn) <b><u>atomic weight = 118</u></b> Oxygen combination: <b>1:2</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Electron affinity: 121</i></p>	<p>Zirconium (Zr) <b><u>atomic weight = 90</u></b> Oxygen combination: <b>1:2</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Electron affinity: 48</i></p>	<p>Cesium (Cs) <b><u>atomic weight = 133</u></b> Oxygen combination: <b>2:1</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Electron affinity: 45</i></p>	<p>Potassium (K) <b><u>atomic weight = 39</u></b> Oxygen combination: <b>2:1</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Electron affinity: 48</i></p>	<p>Cadmium (Cd) <b><u>atomic weight = 112</u></b> Oxygen combination: <b>1:1</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Electron affinity: 0</i></p>
<p>Nitrogen (N) <b><u>atomic weight = 14</u></b> Oxygen combination: <b>2:5</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Melting point: -209</i></p>	<p>Carbon (C) <b><u>atomic weight = 12</u></b> Oxygen combination: <b>1:2</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Electron affinity: 122</i></p>	<p>Aluminum (Al) <b><u>atomic weight = 27</u></b> Oxygen combination: <b>2:3</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Boiling point of salt: 120</i></p>	<p>Silicon (Si) <b><u>atomic weight = 28</u></b> Oxygen combination: <b>1:2</b></p> <p>-----</p> <p><i>Transition Property</i> <i>Electron affinity: 133</i></p>	<p>Undiscovered (Un) <b><u>atomic weight = _____</u></b> Oxygen combination: <b>____</b></p> <p>-----</p> <p><i>Transition Property</i></p>